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have been introduced at specific requests from some of you. Others are still at preparatory stage and will be implemented soon.

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Study Guide and Reinforcement 3
ANSWER KEY 7.

opposes the motion of objects that move

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through the air, is affected by speed, size, and shape 8. net force 9. microwelds 10. rolling 11. air resistance 12. acceleration 13. sliding 14. parachute 1. Gravity is a force that every object in the universe exerts on every other object. 2. their masses ...

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Stoichiometry
Problems If the statement is true, write "true." If it is false, change the underlined word or words to make it true. Write your answer on the line provided.

The major types of stoichiometry problems are mass-mass, mass-volume, and volume-volume.

10-2 Review and Reinforcements - Front Page

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2 Answer.Key 2-1
Explore, page 5 ... 2-1
Review and
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8-9 . 1. energy 2.
radiant 3. chemical
potential 4. joule 5.
created, destroyed 6.
natural gas 7. kinetic ...
11. filtration 12.
filtration 13.
crystallization 14.
distillation 15.
chromatography

Chemistry: Chapter 15 Concentration

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1 BIOLOGY: The

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your textbook, read
about the science of
biology. Answer the
following questions. 1.

What is the primary
focus of all biological
studies? 2.

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Reading the Periodic
Table . On the line at
the left, write the letter

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of the appropriate location of each group of elements on the periodic table below.

- Some letters will be used more than once.
1. carbon family
 8. f-block elements
 2. alkaline earth metals
 9. noble gases
 3. inner transition metals
 10. p-block ...

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3—1 Review and
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Models of the Atom On
the line at the left,
write the letter of the
chemist who proposed
each of the ideas listed
below. Each letter may
be used once, more
than once, or not at all.
a. Democritus b.
Lavoisier c. Proust d.
Dalton 1. Matter is
neither created nor
destroyed in chemical
reactions. 2.

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11—2 Review and Reinforcement Solving Stoichiometry Problems If the statement is true, write "true." If it is false, change the underlined word or words to make it true. Write your answer on the line provided.

1. 2. 5. The major types of stoichiometry problems are mass-mass, mass-volume, and volume-volume.

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a beryllium atom
bonded to 2 chlorine

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atoms. Beryllium has only 4 electrons. Predict the bond angle in this molecule. 6. 8. 9. 10. The molecule NF_3 has nitrogen as the central atom with the 3 fluorine atoms bonded to it. Predict the molecular shape and the bond angles in this molecule. What is the molecular shape of phosphorus trichloride (PCl_3)?

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II—I Review and
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(continued) Solve each
of the following
problems as directed.
Show all your work. 13.
How many moles of
magnesium are
required to react with
2.0 mol of hydrochloric
acid? The equation for
this reaction is $\text{Mg} + 2\text{HCl} \rightarrow \text{MgCl}_2 + \text{H}_2$
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7-2 Review and
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Covalent Bonding. If

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the statement is true, write "true." If it is false, change the underlined word or words to make it true. Write your answer on the line provided.

- ___F, covalent_ 1. A group of atoms united by ionic bonds is called a molecule. _____T_____
2. A covalent bond is formed by a shared pair of ...

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(continued) 9. $0.0621 \text{ mol SS} = (0.0621 \text{ mol} \times 65.8 \text{ g/mol}) = 4.09 \text{ g}$

10. $65.8 \text{ g Mg} = 1. \text{ atoms Mg}$

Solve each of the following problems as directed. Show all your work.

11. Calculate the number of atoms in 0.40 mol of sulfur-

12. Calculate the number of atoms in 2.30 mol of silver.

13. You have a tank of NO_2 gas.

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