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Correct Answer: C. 76.7°C A chemist mixes 75.0 g of an unknown substance at 96.5°C with 1,150 g of water at 25.0°C . If the final temperature of the system is 37.1°C , what is the specific heat capacity of the substance?

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2. Where have you heard the word calorie before?

Student Exploration Calorimetry Lab Answers Activity A

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Calorimetry Lab. Investigate how calorimetry can be used to find relative specific heat values when different substances are mixed with water. Modify initial mass and temperature values to see effects on the system.

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Student Exploration: Seed Germination (ANSWER KEY)

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DOWNLOAD Student Exploration: Calorimetry Lab Vocabulary: calorie, calorimeter, joule, specific heat capacity Prior Knowledge Questions (Do these BEFORE using the Gizmo.) The Latin word

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calor means "heat," and meter comes from the Greek word meaning "to measure." What do you think a calorimeter does?
2. Where have you heard the word calorie before?

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Investigate how calorimetry can be used to find relative specific heat values when different substances are mixed with water. Modify initial mass and temperature values to see effects on the system. One or any combination of the substances can be mixed with water. A dynamic graph (temperature vs. time) shows temperatures of the individual substances after mixing.

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