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Student Exploration: Calorimetry Lab
Vocabulary: calorie, calorimeter, joule, specific heat capacity
Prior Knowledge Questions (Do these BEFORE using the Gizmo.)
1. The Latin word calor means “heat,” and meter comes from the Greek word meaning “to measure.” What do you think a calorimeter does? Measures heat
2. Where have you heard the word

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Gizmo Warm-up A calorimeter is an insulated container filled with a liquid, usually water. When a hot object is

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placed in the calorimeter, heat energy is transferred from the object to the water and the water heats up. Calorimeters can be used to find a substance's specific heat capacity. You will use the Calorimetry Lab Gizmo™ to

Calorimetry Lab Gizmo : Lesson Info : ExploreLearning

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Calorimetry Practice Problems (Answers)

1. How much energy is needed to change the temperature of 50.0 g of water by 15.0°C? 3135J 3140J (rounded answer for sig. figs.) 2. How many grams of water can be heated from 20.0 °C to 75°C using 12500.0 Joules? 119.6 g 120 g (rounded answer for sig. figs) 3.

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different substances are mixed with water. Modify initial mass and temperature values to see effects on the system. One or any combination of the substances can be mixed with water. A dynamic graph (temperature vs. time) shows temperatures of the individual substances after mixing.

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Physics P Worksheet 12.1d Calorimetry Worksheet 12.1d Calorimetry 1. 200 g of water ($C_{\text{water}} = 4180 \text{ J/kg}\cdot\text{K}$) at 60°C is mixed with 200 g of water at 20°C . What is the final temperature of the mixture? 2. 150 g of water at 60°C is mixed with 100g of water at 20°C .

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About This Quiz & Worksheet.

Calorimetry is a complicated science.

This quiz/worksheet will help you assess your understanding of how to calculate temperature and heat capacity and let you put ...

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