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Chapter 17: Reaction Rates. STUDY. PLAY. rate. occurrence per unit of time. speed. example of rate. speed. distance per unit of time. slope. indicate when the reactant rate is increasing more rapidly than the flat areas. slows down. towards the end of the reaction, the reaction rate.

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CHAPTER 17 Reaction Kinetics

560 Chapter 16 • Reaction Rates Section 116.16.1 A Model for Reaction Rates MAIN Idea Collision theory is the key to understanding why some reactions are faster than others. Real-World Reading Link Which is faster: walking to school, or riding in a bus

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Chapter 16 - The Process of Chemical Reactions

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Chapter 16 - The Process of Chemical Reactions 249 Exercise 16.5 - Predicting the Effect of Disruptions on Equilibrium: Nitric acid can be made from the exothermic reaction of nitrogen dioxide gas and water vapor in the presence of a rhodium and platinum catalyst at 700-900 C and 5-8 atm. Predict whether each of the

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Chapter 17: Reaction Rates

Name CHAPTER Date Class STUDY GUIDE FOR CONTENT MASTERY Section 17.2 Factors Affecting Reaction Rates In your textbook, read about the factors that affect reaction rates (reactivity, concentra-