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chapter: Rates and Mechanism of Chemical Reactions (Chapter 22 of Chemical Principles, 3rd Ed (1979) by Dickerson, Gray, and Haight.) Perfectly good for first-year general chemistry courses.

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Chemical change is guided and driven by energetics (thermodynamics), but



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the actual route it takes and the speed with which it occurs is the subject of

"dynamics". Dynamics is itself divided into two general areas: kinetics, which deals with the rate of change and is the subject of this lesson.

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Kinetics and Reaction  
Dynamics, Dover  
Publications.

(Recommended)William

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Francisco, Jeffrey I.

Steinfeld, Chemical

Reaction Dynamics,

Prentice Hall.Raphael

D. Levine, Molecular

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While chemical kinetics  
is concerned with the  
rate of a chemical  
reaction,

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thermodynamics determines the extent to which reactions occur. In a reversible reaction, chemical equilibrium is reached when the rates of the forward and reverse reactions are equal [the principle of dynamic equilibrium] and the concentrations of the reactants and Products no longer change.



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