

Chemistry B Moles Packet Answer Key

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PracticePacket((Unit6: Moles(&Stoichiometry
CHEMISTRY MOLES PACKET PAGE 3 CHEMISTRY WORKSHEET # 1 MOLAR MASS (GRAM MOLECULAR/FORMULA WEIGHTS) We know that grams are actually a measure of the mass of matter and not the weight. Mass is the quantity of matter present; weight is a measure of the pull

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of gravity on matter and is measured in pounds or newtons.

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CHEMISTRY B- MOLES PACKET NAME: _____ HR: _____ PAGE 8 CHEMISTRY WORKSHEET # 5 MOLE PROBLEMS—MOLAR VOLUME OF A GAS We have learned that a mole is a mass of material and number of particles. A mole can also be a measure of volume when we are talking about gases.

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“Molar mass” is used to describe the mass of one mole of a chemical compound, while “atomic mass” is used to describe the mass of one mole of an element or the mass of one atom of an element.

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The molecular formula = --1 B 13. In an experiment, a student gently heated a hydrated $\text{CuSO}_4 \cdot x \text{H}_2\text{O}$ to remove the water. The following data was recorded: Mass of empty crucible Mass of crucible contents before heating Mass of crucible contents after heating Determine the formula of the hydrate. cusc The formula of the hydrate =

CHEMISTRY WORKSHEET # 2 MOLE PROBLEMS—THE MOLE AS A UNIT ...

The mole is a key unit in chemistry. The molar mass of a substance, in grams, is numerically equal to one atom's or molecule's mass in atomic mass units. Exercises

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CHEMISTRY B- MOLES PACKET NAME: _____ HR: _____ PAGE 7CHEMISTRY WORKSHEET # 5 MOLE PROBLEMS—MOLAR VOLUME OF A GASWe have learned two definitions of a mol, now we will learn a third. A mole can also be a measure of volumewhen we are talking about gases.

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Honors Chemistry - Unit 2

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For these elements one mole is 6.02×10^{23} molecules. That is, 6.02×10^{23} molecules of hydrogen is one mole of hydrogen, 6.02×10^{23} molecules of nitrogen is one mole of nitrogen, 6.02×10^{23} molecules of oxygen is one mole of oxygen, etc.

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Pre AP Chemistry Unit 10 HW Packet Name _____ 90% of a worksheet must be completed to earn credit for that worksheet! Page 2 of 10 WKS 10.2 - Calculating with Mass, Moles, & Molecules (1 page) YOU MUST SHOW WORK USING DIMENSIONAL ANALYSIS TO EARN CREDIT FOR THE PROBLEMS! 1. How many grams of ammonia (NH₃)

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