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He received a B.S. Degree in Civil Engineering and an M.S. Degree in Sanitary Engineering from the University of Iowa and a Ph.D. Degree in Environmental Engineering from Stanford University. He taught at Drexel University for eight years before joining the faculty at the University of Iowa in 1986.

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download instant at 2-1 Chemistry for Environmental Engineering and Science, 5th Edition CHAPTER 2 - PROBLEM SOLUTIONS 2.1 (a) $MgCO_3$ FW = 24.3 + 12 + 3(16) = 84.3 g/mol $MgCO_3 + 2H^+ = Mg^{2+} + H_2CO_3$ EW = 84.3/2 = 42.15 g/equiv (b) $NaNO_3$ FW = 23 + 14 + 3(16) = 85 g/mol $NaNO_3 + H^+ = Na^+ + HNO_3$ EW = 85/1 = 85 g/equiv (c) CO_2 FW = 12 + 2(16) = 44 g/mol $CO_2 + H_2O = H_2CO_3$ Note from above: $MgCO_3 + 2H^+ = Mg^{2+} + H_2CO_3$ and $H_2CO_3 = 2H^+ + CO_3^{2-}$ or: $CaCO_3 + 2H^+ = Ca^{2+}$...

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