

## Chemistry If8766 Answers Mixed Mole Problems

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of CO and O2 are required to produce 10 moles of CO2? If 5.0 moles of water is produced, how many moles of hydrogen gas and oxygen gas re required? 21b + O2 2H2O How many moles of tin are produce by reducing 4.35 moles of tin ore with carbon? snO2 + C Sn + CO2 How many moles of hydrogen are needed to completely react with two moles of nitrogen?

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Stoichiometry: Mixed Problems (KEY) 1) N2 + 3H2 → 2NH3 What volume of NH3 at STP is produced if 25.0 of N2 is reacted with an excess of H2? 3 3 3 2 2 2 40.0L NH 1mol NH 22.4L NH 1mol N 2mol NH 28.0g N 25.0g N 1mol N × × × = 2) 2KClO3 → 2KCl + 3O2 If 5.0g of KClO3 is decomposed, what volume of O2 is produced at STP? 2

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MIXED MOLE PROBLEMS - KEY 1. a) How many grams are there in 1.5 × 1025 molecules of CO 2? 1.110g 1 mol 44.0 g 6.0210molecules 1.510molecules CO1mol 3 23 2 251=!!!! b) What volume would the CO

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Answer Key Chemistry If8766 Moles And Mass Download or Read Online eBook answer key chemistry if8766 moles and mass in PDF Format From The Best User Guide Database Moles and Mass. Directions: Determine the number of moles in each of the quantities below. 1) 52 g of NaCl. ANSWER: 0.89 mole Chemistry If8766.

**The Mole And Volume Worksheet Answers Instructional Fair Inc**  
ANSWER: 9.3 E 21 atoms 6) A 4.0g sample of O 2 is in a container at STP. What volume is the container? ANSWER: 2.8 L 7) How many molecules of O 2 are in the container in Problem 6? How many atoms of oxygen are in the container? ANSWER: 7.5 E 22 molecules Chemistry If8766 Page 53 Instructional Fair, Inc.

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How many moles of oxygen are necessary to react completely with four moles of propane (C3He)? 5. + 3KN03 + How many moles of potassium nitrate are produced when two moles of potassium phosphate react with moles of aluminum nitrate? enstructional Fair, Inc. Chemistry If8766 62

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