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4 74 g of KCl z LB (03 a9/)3c 168 , 0991957/0/ m l.t9eoS D enstructional Fair, Inc. 5. 35 g of  $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$  Determine the number of grams in each of the quantities below. 1, 2.5 moles of NaCl 2. 0.50 moles of 3, 1.70 moles of  $\text{KMnO}_4$  4, 0.25 moles of KCl e, ST 5. 3.2 moles of  $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$  Chemistry IF8766

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Writing Formulas From Names Directions: Write the formulas of the following compounds. 1. ammonium phosphate (NH<sub>4</sub>)<sub>3</sub> PO<sub>4</sub> \_\_\_\_ 2. iron (II) oxide FeO \_\_\_\_ 3. iron (III) oxide Fe<sub>2</sub> O<sub>3</sub> ... Chemistry IF8766 Page 48 Instructional Fair, Inc. Title: Name:

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**SIGNIFICANT FIGURES Name**

r i c Page 4:2 ANSWER KEY WRITING FORMULAS (CRISS-CROSS METHOD) Name IONIC BONDING Name \_ IonicbondingoccurswhenametcI hcnstersonemoreelectrons10anonmelolln on ...

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Chemistry IF8766 9 OInstructional Fair, Inc. 59. CALCULATIONS USING SIGNIFICANT FIGURES ... The answer is expressed as 190,000 cm<sup>3</sup> since 19 cm has only two significant fiaures. | | When adding and subtracting, limit and round your answer to the least number of decimal places in any of the numbers that make up your answer.

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