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Lesson 2.2: Physical Science – Atoms and Molecules H. Turngren, Minnesota Literacy Council, 2014 p.9 GED Science Curriculum SCIENCE Unit 2.2 Handout 3 TEACHER ANSWER KEY 1. B 2. C 3. Answers may vary: Isotopes have more neutrons than a stable atom of the same element. 4. 146 5.

Physical Science answers - Answers

substance. All chemical changes include physical changes. Classify the following properties as either chemical or physical by putting a check in the appropriate column. Physical Chemical Property Property 1. red color 2. density / 3. flammability 4. solubility 5. reacts with acid to form hydrogen 6. supports combustion / 7. bitter taste 8. melting point

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Best Answer: Alright basically, when you are balancing an equation, you want both sides to have the same amount of elements. The law states that matter cannot be created nor destroyed. So you can't create new elements and destroy any. So we'll count how many elements are in each side. Clearly, they are not balanced.

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Lesson 2.2: Physical Science Atoms and Molecules

Solution: Use the following relationships: 2, 3, 4. 1000 mg = 1 g kg Start with the original number and unit. Multiply by a unit factor with the unit to be discarded on the bottom and the desired unit on top. Cancel units. Perform numerical calculations.

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ACTIVITY OF THE ELEMENTS Name Since metals prefer to give away electrons during chemical bonding, the most active metals are closest to francium, which is a large atom with low ionization energy and electronegativity, Nonmetals prefer to pull in electrons, so the most active nonmetals are closest to fluorine, which has a high ionization energy and ...