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internal pressure data given in Model 1. Read This! ... Gas Variables 7 Extension Questions 22. Draw a sample of gas that is colder than all three of the samples in Question 21.

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Chemists combine all of the relationships seen in Models 1 and 2 into one law—the Ideal Gas Law. It is one equation that describes gas behavior and the relationship among all four variables, P, V, T, and n. In the Ideal Gas Law the proportionality constant is represented by the letter R (rather than the generic k). 24.

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GAS VARIABLES POGIL Proportionality
Directly Proportional As \square one increases, so does the other = $\square \times \square$
Inversely Proportional As \square one increases, the other decreases = $\square \div \square$
Thermodynamic Observables Aspects of a system that we can observe, measure, or calculate in some way.
Pressure - P kilopascals (kPa)
Temperature - T Kelvin (K)
Volume - V Liters (L)
moles - n ...

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C Adin more as (g g) 5 2 3 W? 5 C1 C2
Volume = 1

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