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Hidrogen peroksida adalah senyawa kimia dengan rumus H 2 O 2.Dalam bentuk murninya, ia berupa cairan bening berwarna biru pucat, sedikit lebih kental daripada air.Hidrogen peroksida adalah peroksida paling sederhana (senyawa dengan ikatan tunggal oksigen–oksigen). H 2 O 2 digunakan sebagai oksidator, zat pemutih, dan antiseptik.Hidrogen peroksida pekat, atau "peroksida uji tinggi", adalah ...

Hidrogen peroksida - Wikipedia bahasa Indonesia ...
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Solutions For Lehninger Biochemistry Ch
Answer Multiplying volume (L) by molar concentration (mol/L) gives the number of moles in that volume of solution. If x is the concentration of gastric HCl (mol/L), (0.010 L)x x Given that pH

Proteins - SlideShare
An acid is a molecule or ion capable of either donating a proton (i.e., hydrogen ion, H +), known as a Brønsted–Lowry acid, or, capable of forming a covalent bond with an electron pair, known as a Lewis acid.. The first category of acids are the proton donors, or Brønsted–Lowry acids.In the special case of aqueous solutions, proton donors form the hydronium ion H 3 O + and are known as ...

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Chemical Principles: The Quest for Insight: Atkins, Peter ...
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The most common crown ethers are oligomers of ethylene oxide with repeating units of (—CH 2 CH 2 O—) n where n = 4 (tetramer), n = 5 (pentamer), or n = 6 (hexamer). Crown ethers are given structural names, X -crown- Y , where X is the total number of atoms in the ring and Y is the number of these atoms that are oxygen.

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