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### The Journal of Physical Chemistry Letters | Vol 12, No 12

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### Calculus Volume 1 - Open Textbook Library

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### History of chemistry - Wikipedia

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### Bond Energies - Chemistry LibreTexts

Chemistry 2e is designed to meet the scope and sequence requirements of the two-semester general chemistry course. The textbook provides an important opportunity for students to learn the core concepts of chemistry and understand how those concepts apply to their lives and the world around them.

### Gas Laws: Overview - Chemistry LibreTexts

The average bond energies in Table T3 are the averages of bond dissociation energies. For example the average bond energy of O-H in H<sub>2</sub>O is 464 kJ/mol. This is due to the fact that the H-OH bond requires 498.7 kJ/mol to dissociate, while the O-H bond needs 428 kJ/mol.

### Textbook Physical Chemistry Vol 3

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